WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau



INTERNATIONAL APPLICATION PUBLISI	HED U	UNDER THE PATENT COOPERATION TREATY (PCT)
(51) International Patent Classification 6:		(11) International Publication Number: WO 97/4117
C08K 5/521, 5/3492, C08L 23/10	A1	(43) International Publication Date: 6 November 1997 (06.11.9)
(21) International Application Number: PCT/EP		DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE).
(22) International Filing Date: 24 April 1997 (2) (30) Priority Data:	24.04.9	Published With international search report.
08/638,417 26 April 1996 (26.04.96)	τ	Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt amendments.
(71) Applicant: AKZO NOBEL N.V. [NL/NL]; Velperweg 6824 BM Amhem (NL).	; 76, N	L-
(72) Inventors: MOY, Paul, Y.; 10 Sandi Lane, Fishkill, N (US). TELSCHOW, Jeffrey, E.; 416 Benedict Tarrytown, NY 10591 (US). ALESSIO, Gerald, Lincoln Boulevard, Emerson, NY 07630 (US).	Avenu	ie,
(74) Agent: SCHALKWIJK, Pieter, Cornelis; Akzo Not Patent Dept. (Dept. APTA), P.O. Box 9300, NL- Amhem (NL).		· 1

(54) Title: FLAME RETARDANT COMPOSITION FOR POLYMERS

(57) Abstract

The present invention is a flame retardant composition, adapted to be mixed with a polymer substrate to confer flame retardancy on the substrate, which comprises: (a) a bicyclic phosphorus flame retardant compound, such as one containing one or more pentaerythritol phosphate alcohol moieties, as exemplified by bis(pentaerythritol) phosphate alcohol) carbonate; (b) an intumescent flame retardant compound containing nitrogen and phosphorus, such as melamine phosphate; and (c) a monophosphate ester compound to enhance the charring and processing characteristics of the composition in the polymer substrate, such as a liquid aryl-group containing phosphate ester compound, and the monophosphate ester compound is triphenyl phosphate.

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	∧lbania	ES	Spain	LS	1.esotho	SI	Slovenia	
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia	
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal	
ΑU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland	- 1
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad	
BA	Bosnia and Herzegovina	GR	Georgia	MD	Republic of Moldova	TG	Togo	
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan	
BE	Belgium	GN	Guinea	MK	The former Yugoslav	TM	Turkmenistan	
BF	Burkina Faso	GR	Greece		Republic of Macedonia	TR	Turkey	Ì
BG	Bulgaria	HU	Hungary	ML	Mali	1 T	Trinidad and Tobago	
ВJ	Benin	1E	Ireland	MN	Mongolia	UA	Ukraine	
BR	Brazil	IL.	Israel	MR	Mauritania	UG	Uganda	
BY	Belarus	IS	Iceland	MW	Malawi	บร	United States of America	- 1
CA	Canada	ıT	Italy	MX	Mexico	UZ	Uzbekistan	ı
CF	Central African Republic	JP	Japan	NE	Niger	VN	Viet Nam	
CG	Congo	KE	Кепуа	NL	Netherlands	YU	Yugoslavia	
CH	Switzerland	KG	Kyrgyzstan	NO	Norway	zw	Zimbabwe	ļ
CI	Côte d'Ivoire	KP	Democratic People's	NZ	New Zcaland			
CM	Cameroon		Republic of Korea	PL	Poland			
CN	China	KR	Republic of Korea	PT	Portugal			- 1
CU	Cuba	KZ	Kazakstan	RO	Romania			
CZ	Czech Republic	LC	Saint Lucia	RU	Russian Federation			
DE	Germany	LI	Liechtenstein	SD	Sudan			
DK	Denmark	LK	Sri Lanka	SE	Sweden			1
EE	Estonia	LR	Liberia	SG	Singapore			
								1

WO 97/41173 PCT/EP97/02159

FLAME RETARDANT COMPOSITION FOR POLYMERS

5 Background of the Invention

10

15

Flame retardant compositions comprising certain bicyclic phosphorus flame retardant compounds and certain intumescent flame retardant compounds which are suitable for use in a variety of polymers, including polyolefins, are disclosed in U.S. Patent No. 4,801,625 to W.J. Parr et al. While such compositions are useful as flame retardants for such application, persons of ordinary skill in the art have desired improved flame retardant compositions which have enhanced processing characteristics with the selected polymer they are intended to flame retard. Processing improvements include less porous pellets and/or a smoother surface characteristic when extruded. The present invention is directed to addressing these needs.

Summary of the Invention

20 The present invention relates to a flame retardant composition, adapted to being mixed with a polymer substrate to confer flame retardancy on the substrate, which comprises: (a) a bicyclic phosphorus flame retardant compound; (b) an intumescent flame retardant compound containing nitrogen and phosphorus; and (c) a monophosphate ester compound to enhance the 25 compatibility of the composition in the polymer substrate. The bicyclic phosphorus flame retardant compound can be a compound containing one or more pentaerythritol phosphate alcohol moieties, such as bis(pentaerythritol phosphate alcohol) carbonate. The intumescent flame retardant compound containing nitrogen and phosphorus can be melamine phosphate. The 30 monophosphate ester compound can be a liquid aryl-group containing phosphate ester compound, such as isodecyl diphenyl phosphate, or can be a

WO 97/41173 PCT/EP97/02159

2

solid aryl-group containing phosphate ester compound, such as triphenyl phosphate.

Description of the Preferred Embodiments

5

The present invention's essential components are, preferably, as described herein below.

The Bicyclic Phosphorus Flame Retardant

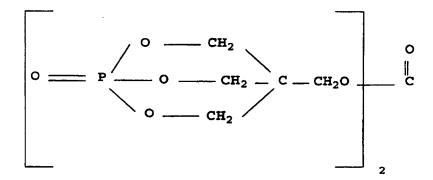
10

20

Included within the definition of possible bicyclic phosphorus flame retardant compounds, which are solid materials, for use as one component of the flame retardant composition of the present invention are the following representative compounds:

Pentaerythritol phosphate alcohol itself, which forms a reagent to manufacture a number of the compounds to be described hereinafter, is one flame retardant which can be selected for use in connection with the present invention.

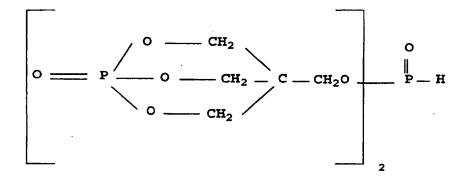
Bicyclic phosphate ether, ester, and carbonate flame retardants as described in U.S. Patent No. 4,801,625 to W.J. Parr et al., which is incorporated herein by reference, can be employed and form a preferred class of such component(s) herein. A preferred species is bis(pentaerythritol phosphate alcohol) carbonate which is of the formula:



Bis(pentaerythritol phosphate alcohol) hydrogen phosphonate may be used as the pentaerythritol phosphate alcohol-derived flame retardant additive in accordance with the present invention, and it is of the following formula and is described in U.S. Patent No. 5,420,326:

5

20



One process for forming the above-described type of compound is by the transesterification of a diphenyl phosphite with pentaerythritol phosphate alcohol which has the formula:

The phosphite reagent is of the formula (ArO)₂P(O)H, where Ar is substituted or unsubstituted phenyl. The transesterification

reaction is advantageously conducted at elevated temperature (e.g., a temperature of from about 120°C to about 250°C, preferably in a high boiling organic solvent, such as an aryl phosphate solvent (as described in U.S. Patent No. 5,237,085), using an appropriate transesterification catalyst (e.g., magnesium dichloride, sodium phenoxide, or the like). The reaction mixture will contain the desired crude product with a phenolic by-product. The crude product can be triturated with a solvent such as acetonitrile or

methanol to give the desired, purified product.

5

10

15

20

25

Bis(pentaerythritol phosphate alcohol) alkylphosphonate

compounds can also be used in the present invention as the pentaerythritol phosphate alcohol-derived flame retardant component. These compounds, which are described in U.S. Patent No. 5,362,898, are of the formula:

where R is alkyl, for example alkyl of from one to four carbon atoms, preferably: methyl. One process for forming the above described compounds is by the transesterification of a diphenyl alkyl phosphonate carrying the desired alkyl group with pentaerythritol phosphate alcohol which has been previously depicted above. The phosphonate reagent is of the formula (ArO)₂P(O)R, where Ar is substituted or unsubstituted phenyl and R is alkyl as previously described. The transesterification reaction is advantageously conducted at elevated temperature (e.g., a temperature of from about 170°C to about 200°C) in a high boiling organic solvent, such as an aryl phosphate solvent (as described in U.S. Patent No. 5,237,085), using an appropriate transesterification catalyst (e.g., magnesium dichloride, sodium phenoxide, or the like). The reaction mixture will contain the desired crude product with a phenolic by-product. The crude product can be triturated with a solvent such as acetonitrile or methanol to give the desired, purified product. An alternative way of preparing these compounds is by the reaction of pentaerythritol phosphate alcohol, a trialkylamine, such as triethylamine, and an alkylphosphonic dihalide, such as methylphosphonic dichloride, in an appropriate solvent, such as

acetonitrile, under cooling at essentially ambient temperature (e.g., 20°C to about 30°C).

Another type of compound that can be used has the following formula and is bis(pentaerythritol phosphate alcohol) pentaerythritol phosphate alcohol phosphonate and has the following formula:

where R in the above formula is derived from pentaerythritol phosphate alcohol and is of the formula:

This compound can be formed by a known pentaerythritol phosphate alcoholtriaryl phosphite transesterification reaction followed by Arbuzov rearrangement.

This invention also contemplates use, as the pentaerythritol phosphate alcoholderived flame retardant, of the (pentaerythritol phosphate alcohol) (cyclic neopentyl glycol) phosphite compound of the formula:

and the (pentaerythritol phosphate alcohol) (cyclic neopentyl glycol)

5 phosphonate compound of the formula:

One process for forming the above-described phosphite compound is by the reaction of a neopentylene glycol halophosphite with pentaerythritol phosphate alcohol which has been previously depicted hereinabove. The neopentylene glycol halophosphite reagent is of the formula:

15

20

where X is a halogen atom, such as chlorine. The reaction can be conducted at room temperature in an appropriate organic solvent, such as acetonitrile using an acid acceptor, such as a trialkylamine. Once the phosphite compound has been synthesized, it can be converted to the phosphonate compound of the present invention by heating, for example, in a high boiling organic solvent, such as an aryl phosphate solvent.

30

The above described type of flame retardant component can be used in the present invention of from about 1 % to about 30 %, by weight of the composition, preferably from about 5 % to 12 %.

5 The Intumescent Flame Retardant

The type of intumescent flame retardant component which also forms a part of the present invention includes the intumescent phosphorus acid salts of amines or ammonia. They contain both phosphorus and nitrogen atoms. Included within this category of useful reagents for use herein are such phosphates as melamine phosphate (which is preferred), ethylenediamine phosphate, and cyanoguanidine phosphate, such polyphosphates as ammonium polyphosphate, and such pyrophosphates as ammonium pyrophosphate.

This type of intumescent flame retardant additive can be used in amounts ranging from about 5% to about 30 % by weight of the composition, preferably from about 15% to about 20%.

The Monophosphate Ester

The novel additive for the flame retardant composition of the present invention as compared to U.S. Patent No. 4,801,625 to W.J. Parr et al. is a monophosphate ester compound to enhance the compatibility of the known two-component flame retardant in the selected polymer to be flame retarded. Included as useful additives of this type are the aryl group-containing monophosphate esters which can be liquid (such as the alkyl diaryl phosphates, as exemplified by isodecyl diphenyl phosphate), or solid (such as triphenyl phosphate).

The amount of this monophosphate additive which can be used in the compositions of the present invention can range from about 1% to about 30%, by weight of the composition, preferably from about 5% to about 15%.

The compositions of the present invention can be formed by mixing the foregoing additives in such conventional mixing equipment as: single or multiple screw extruders; internal mixers, such as the Banbury mixer; or other melt blending plastic compounding devices.

- 5 The type of polymers to which the foregoing type of flame retardant can be added include the polyolefins, the polyamides, the polycarbonates, the polyacrylates, the polyesters, and the polyphenylene oxides. The weight amount of such flame retardant in such polymer can, in general, range from about 10% to about 30%, by weight of the polymer.
- In addition to the flame retardant composition of the present invention, the selected polymer to be flame retarded can include one or more other functional additives known to persons of ordinary skill in the art of polymer compounding and processing, including fillers, antioxidants, lubricants, stabilizers, pigments, rheology modifiers, impact modifiers, and antistatic agents.

15

The present invention is further illustrated by the Examples which follow.

EXAMPLES

5

10

15

20

25

30

The Examples which follow demonstrate that bicyclic phosphorus flame retardant materials in polypropylene (PP) composites have shown surprising significance to low level additions (less than or equal to about 2.5%) of a phosphate ester plasticizer (for example, isopropylphenyl diphenyl phosphate). Char forming effectiveness was enhanced an additional ten percent (as seen in cone calorimetry evaluations: ASTM-1354-90). Other flammability tests such as the UL-94/V-0 protocol, showed marked improvement from a dripping V-2 rating up to V-0 with composites containing the phosphate ester plasticizer (see Tables 1 and 2). This behavior was especially striking in the bis(pentaerythritol phosphate alcohol) pentaerythritol phosphate alcohol phosphonate additive ("BP³" in the Tables given below. The abbreviation "PEPA" standing for pentaerythritol phosphate alcohol). Combined with up to twenty five percent (by weight) of the isopropylphenyl diphenyl phosphate, the FR effectiveness was improved to the higher levels of performance.

The addition of the plasticizer improved the ductility and the tensile elongation of the composite. Also improved was the dispersion of additives in the composites. Surprisingly, plasticizer migration was not observed despite exposure to accelerated testing conditions (150° C) for over thirty days.

The Tables in the following Examples demonstrate the physical properties and flammability performance.

A series of flammability trials were performed, examining the effects of low level additions of an isopropylphenyl diphenyl phosphate, which is also termed "isopropylated phosphate" or "isopropyl diphenyl phosphate" in the tables which follow. Initial trials of bis(PEPA) PEPA phosphonate in polypropylene looked at the variation of FR adjuvants:

Table 1

Formulations:	#1	#2	#3	#4	#5	#6
PP-6323, polypropylene (Himont)	66.2	68.2	71.2	68.2	66.2	61.2
BP ³ / isopropylated phosphate (< 5%)	13.5	11.5	11.5	13.5	11.5	11.5
Melamine Phosphate	18	18	15	15	21	21
THEIC*	2	2	2	3	1	3
Calcium Stearate	0.3	0.3	0.3	0.3	0.3	0.3
UL-94 (1/16")	Fail	Fail	Fail	Fail	Fail	Fail

^{*}tris (hydroxyethyl) isocyanurate

Despite the high loading levels and significant amounts of phosphorus (up to about 2.9%) contributed from the intumescent flame retardant, little to no FR efficacy was seen.

Several levels of the isopropylphenyl diphenyl phosphate (PHOSFLEX 41P brand) were incorporated into BP³ and were compounded again into polypropylene with the same adjuvants with the following results:

11

Table 2

Formulations:	#1	#2	#3	#4
PP-6323, polypropylene (Himont)	66.2	68.2	71.2	68.2
BP ³ /isopropylated phosphate (about 15%)	13.5	-	13.5	-
BP ³ /isopropylated phosphate (about 25%)	<u>-</u>	13.5	-	13.5
Melamine Phosphate	18	18	18	18
THEIC	2	2	2	3
Calcium Stearate	0.3	0.3	0.3	0.3
UL-94 (1/16") Evaluation #1	V0/	V2	-	-
	V1			
UL-94 (1/16") Evaluation #2	-	-	V0	V0

Tested in two concurrent evaluations, both studies showed marked flame retardant improvement over non-plasticized composites.

For further testing of the flame resistant effect from varying plasticizer dosage levels, the following formulations were prepared and evaluated in a cone calorimeter apparatus (ASTM 1354-90):

Table #3

Formulations:	#1	#2	#3	#	#5
PP-6323 (Himont)	8.69	8.69	8.69	8.69	8.69
(BPC) ¹	10		,	•	
(BPC-15% 41P) ²	1	10	ı	-	,
(BPC-25% 41P) ³	1	t	1	10	•
(BP ³ -18% 41P) ⁴	3	\$	10	1	•
(BP³-25% 41P) ⁵	•	•	•	1	10
Melamine Phosphate	18	18	18	18	18
Tris hydroxyethyl isocyanurate	2	2	2	2	2
Calcium Stearate	0.2	0.2	0.2	0.2	0.2

1) bis (PEPA) carbonate

bis (PEPA) carbonate with 15 % isopropyl diphenyl phosphate (Phosflex® 41P)
 bis (PEPA) carbonate with 25 % isopropyl diphenyl phosphate (Phosflex® 41P)
 bis (PEPA) PEPA phosphonate with 18 % isopropyl diphenyl phosphate (Phosflex® 41F)
 bis (PEPA) PEPA phosphonate with 25 % isopropyl diphenyl phosphate (Phosflex® 41P)

Total Heat (MJ/m²)269 198 194 209 199 207 Effect. Heat of Combust (MJ/kg) (300 sec) 34.0 22.3 20.0 25.3 30.2 23.1 Heat Release Rate (kW at Sec) 1705 / 160 340 / 176 383 / 176 394 / 179 331 / 172 363 / 167 Peak Avg. Peak Heat Release (kW at Sec) 484 / 180 157 / 373 159 / 373 168 / 365 186 / 349 173/381 Rate Cone Calorimetry Evaluations of IFR/PP Composites Spiked with IP4 ō Release Rate (kW/m²) Avg. Heat 125 297 128 132 137 141 Mass Loss (%) 82.5 71.9 53.2 50.2 81.1 ٧ Time of Ignit. (sec.) 110 119 109 131 58 98 Heat Flux Rate KW/m² 35 35 35 35 35 35 PP-6323 (Himont) BPC/15% - 41P BPC/25% - 41P BP³/18% - 41P BP³/ 5% - 41P BPC, neat Materials:

UL-94 and LOI Flammability Performance of IFR/PPs	rformance of IFR/P	Ps	
Materials:	UL-94 (1/16")	UL-94 (1/8")	LOI (0 ₂ %)
PP-6323, Polypropylene - Himont	•	•	ı
BPC, Neat	۸-0	٨-0	33.2
(BPC-15% 41P)	٨-0	٨-0	33.4
(BPC-25% 41P)	۸-0	0-/	32.2
(BP³-18% 41P)	۸-0	0-/	34.3
(BP ³ -25% 41P)	0-/	٨-0	33.5

WO 97/41173 PCT/EP97/02159

Physical Property Performance of IFR/PPs

10

15

20

Material:	Ten. Yield Str. (10 ³ psi)	Ten. Elong %	Flex. Str. 10 ³ psi	Flex. Mod. 10 ⁵ psi	Izod Imp. ft-Ib/in	HDT 264 °C psi	MFI Cond. L
PP-6323, Polypropylene - Himont	4.2	488	3.6	1.6	0.3	53.2	8.4
BPC, Neat	3.6	6	5.2	2.9	0.3	59.1	8.3
(BPC-15% 41P)	3.3	11	4.7	2.5	0.3	56.1	9.5
(BPC-25% 41P)	3.4	19	4.6	2.4	0.4	54.1	9.5
(BP ³ -18% 41P)	3.3	13	4.9	2.6	0.3	69.2	6.5
(BP ³ -25% 41P)	3.2	23	4.3	2.4	0.3	60.2	8.2

5 In the preceding Tables the following conclusions can be drawn:

All samples shown in the table displaying the cone calorimetry data exhibited exceptional flame retardancy characteristics over virgin polypropylene. Of particular note is the mass loss (%) of these composites. A lower number implies a greater degree of char formation (namely, less mass loss). Of the three BPC-containing composites, the mass loss decreased as the dosage of the plasticizer increased. In effect, a one percent increase of plasticizer in the overall composite appeared to be responsible for a ten percent increase in char formation.

In the table showing the UL-94 and LOI data, all listed intumescent fire retardant composites demonstrated excellent LOI (ASTM 2863) and UL flammability results.

In the Table illustrating the physical property performance data, it was observed that, despite the presence of plasticizer in a number of the formulations, its effect upon the heat deflection results was minimal. All plasticizer-containing composites showed better dimensional stability than the neat polypropylene resin (PP-6323).

In the Table listed below, three additional composites were fabricated and tested in which the bis(PEPA) carbonate ("BPC") was combined with three differing phosphate ester additives:

Table

5

Formulations:	#1	#2	#3
PP-6323 (Himont)	69.7	69.7	69.7
BPC ¹ and 2-ethylhexyl diphenyl phosphate	10	_	-
BPC ² and t-butylphenyl diphenyl phosphate	-	10	-
BPC ³ and triphenyl phosphate	-	-	10
Melamine Phosphate	18	18	18
Calcium Stearate	0.3	0.3	0.3
UL-94 (1/16")	V-O	V-0/	V-O/
		V-1	V-2

The above data indicate that the alkylated diphenyl phosphate-containing composite exhibited a better degree of flame retardancy than the other two composites that were tested.

10

The forgoing Examples, which have been set forth for illustrative purposes only, should not be construed in a limiting sense for that reason. The scope of protection sought is set forth in the Claims which follow.

PCT/EP97/02159

1. A flame retardant composition, adapted to be mixed with a polymer substrate to confer flame retardancy on the substrate, which comprises: (a) a bicyclic, pentaerythritol phosphate alcohol-derived phosphorus flame retardant compound; (b) an intumescent flame retardant compound containing nitrogen and phosphorus; and (c) an aryl group-containing monophosphate ester compound to enhance the processing of the composition in the polymer substrate.

10

25

- 2. A flame retardant composition as claimed in Claim 1 wherein the bicyclic phosphorus flame retardant compound is a compound containing one or more pentaerythritol phosphate alcohol moieties.
- 15 3. A flame retardant composition as claimed in Claim 1 wherein the bicyclic phosphorus flame retardant compound is bis(pentaerythritol phosphate alcohol) carbonate.
- 4. A flame retardant composition as claimed in Claim 1 wherein the intumescent flame retardant compound containing nitrogen and phosphorus is melamine phosphate.
 - 5. A flame retardant composition as claimed in Claim 1 wherein the bicyclic phosphorus flame retardant compound is a compound containing one or more pentaerythritol phosphate alcohol moieties and the intumescent flame retardant compound containing nitrogen and phosphorus is melamine phosphate.
- 6. A flame retardant composition as claimed in Claim 1 wherein the bicyclic phosphorus flame retardant compound is bis(pentaerythritol phosphate

PCT/EP97/02159

alcohol) carbonate and the intumescent flame retardant compound containing nitrogen and phosphorus is melamine phosphate.

- 7. A flame retardant composition as claimed in Claim 1 wherein the
 5 monophosphate ester compound is a liquid aryl-group containing phosphate ester compound.
 - 8. A flame retardant composition as claimed in Claim 1 wherein the bicyclic phosphorus flame retardant compound is a compound containing one or more pentaerythritol phosphate alcohol moieties, the intumescent flame retardant compound containing nitrogen and phosphorus is melamine phosphate, and the monophosphate ester compound is a liquid aryl-group containing phosphate ester compound.
- 9. A flame retardant composition as claimed in Claim 1 wherein the bicyclic phosphorus flame retardant compound is bis(pentaerythritol phosphate alcohol) carbonate, the intumescent flame retardant compound containing nitrogen and phosphorus is melamine phosphate, and the monophosphate ester compound is a liquid aryl-group containing phosphate ester compound.

20

- 10.A flame retardant composition as claimed in Claim 1 wherein the monophosphate ester compound is triphenyl phosphate.
- 11.A flame retardant composition as claimed in Claim 1 wherein the bicyclic phosphorus flame retardant compound is a compound containing one or more pentaerythritol phosphate alcohol moieties, the intumescent flame retardant compound containing nitrogen and phosphorus is melamine phosphate, and the monophosphate ester compound is triphenyl phosphate.

WO 97/41173 PCT/EP97/02159

19

12.A flame retardant composition as claimed in Claim 1 wherein the bicyclic phosphorus flame retardant compound is bis(pentaerythritol phosphate alcohol) carbonate, the intumescent flame retardant compound containing nitrogen and phosphorus is melamine phosphate, and the monophosphate ester compound is triphenyl phosphate.

INTERNATIONAL SEARCH REPORT

Inte: anal Application No PCT/EP 97/02159

A. Cl.ASS IPC 6	C08K5/521 C08K5/3492 C08L23/	10			
According	to International Patent Classification (IPC) or to both national class	nfication and IPC			
	S SEARCHED				
Minimum o	locumentation searched (classification system followed by classification contains the context of the classification system followed by classification context of the context of the classification system followed by classification context of the conte	ation symbols)			
Documenta	tion searched other than minimum documentation to the extent that	such documents are included in the fields s	searched		
Electronic o	lata base consulted during the international search (name of data ha	ase and, where practical, search terms used)			
C. DOCUM	MENTS CONSIDERED TO BE RELEVANT				
Category *	Citation of document, with indication, where appropriate, of the	relevant passages	Relevant to claim No.		
A	1983				
see claims 1-5; examples 1,4,43; table 1					
Α	EP 0 305 002 A (AKZO) 1 March 1989 cited in the application see page 2, line 39 - line 41 see page 4, line 4 - line 5; claims 1,8-10; examples 36-43				
Α	WO 91 04294 A (GREAT LAKES) 4 Apssee page 8, paragraph 3 see page 9, paragraph 1; claims example 2		1,7		
Furt	ner documents are listed in the continuation of box C.	X Patent family members are listed in	n annex.		
'A' docume consider filing d'L' docume which i citation 'O' docume other n 'P' docume later th	int which may throw doubts on priority claim(s) or is cited to establish the publication date of another in other special reason (as specified) interested in the control of the control o	"T" later document published after the interest of priority date and not in conflict with cited to understand the principle or the invention." "X" document of particular relevance; the cannot be considered novel or cannot involve an inventive step when the document of particular relevance; the cannot be considered to involve an involve an involve and the cannot be considered to involve an involve and the cannot be considered to involve an involve and the cannot be considered to involve an involve and the cannot be combined with one or ments, such combination being obvious in the art. "&" document member of the same patent."	th the application but ecry underlying the claimed invention be considered to cument is taken alone claimed invention ventive step when the core other such docusts to a person skilled family		
27	7 August 1997				
	nailing address of the ISA European Patent Office, P.B. 5818 Patentiaan 2 NL - 2280 HV Rijswijk Tel. (+ 31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+ 31-70) 340-3016	0 5. 09. 9 Authorized officer Engel, S	37		

INTERNATIONAL SEARCH REPORT

information on patent family members

Interr nal Application No PCT/EP 97/02159

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 69500 A	12-01-83	US 4341694 A CA 1176000 A JP 1779491 C JP 4057707 B JP 58017126 A	27-07-82 09-10-84 13-08-93 14-09-92 01-02-83
EP 305002 A	01-03-89	US 4801625 A CA 1307613 A DE 3851812 D DE 3851812 T	31-01-89 15-09-92 17-11-94 11-05-95
WO 9104294 A	04-04-91	AU 6625390 A AU 7035594 A CA 2025172 A EP 0500608 A IL 95673 A JP 5500983 T	18-04-91 03-11-94 16-03-91 02-09-92 26-05-95 25-02-93